

PHLEGMACINS AND ANHYDROPHLEGMACINQUINONES: DIMERIC HYDROANTHRACENES FROM SEEDLINGS OF *CASSIA TOROSA**

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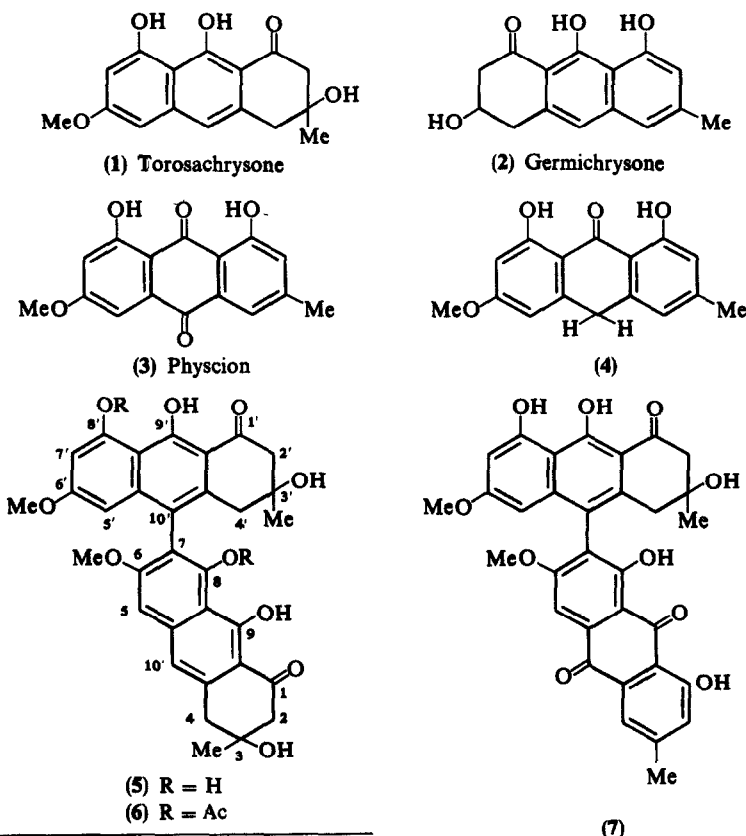
Key Word Index—*Cassia torosa*; Leguminosae; phlegmacin; anhydrophlegmacin-9,10-quinone; hydroanthracene; atropisomer.

Abstract—From seedlings of *Cassia torosa* four dimeric hydroanthracenes have been isolated. Two, a pair of atropisomeric dimers consisting of two molecules of torosachryson, were identified as phlegmacin A_2 and B_2 , the enantiomers of phlegmacin A_1 and B_1 obtained from a higher fungus, *Cortinarius odorifer*. The other pair of atropisomers, based on torosachryson and physcion, were identified as anhydrophlegmacin-9,10-quinones A_2 and B_2 .

INTRODUCTION

Previously we reported the isolation of several anthraquinones and a hydroanthracene, torosachryson (1), from the seeds of *Cassia torosa* Cavanilles [1]. Recently, germichryson (2), another hydroanthracene, was isolated as a characteristic constituent of the seedlings of this plant [2]. During these studies four dimeric hydroanthra-

cen, a pair of atropisomeric dimers consisting of two molecules of torosachryson (1) and the other pair of atropisomeric dimers consisting of torosachryson (1) and physcion (3) have been isolated from the seedlings. This paper describes the isolation and the identification of these dimeric compounds, and also discusses their stereochemistry and biogenesis.



* Part 10 in the series 'Studies of the Constituents of Purgative Crude Drugs'. For part 9 see ref. [2].

RESULTS AND DISCUSSION

Dimeric hydroanthracenes, H and I, obtained from polar fractions of column chromatography of C_6H_6 extracts showed different R_f values on TLC and completely opposite ORD curves, whereas their UV, IR and MS were almost identical to each other. In the PMR spectra of compounds H and I, significant differences were observed in the signals due to $-CH_2-$ groups, while the other signals were practically identical. The UV absorption showed the presence of hydroanthracene chromophore [1, 2] and the MS showed molecular ions at m/e 574 (M^+ ; $C_{32}H_{30}O_{10}$), which corresponded to those of the dimers of torosachryson (1). They showed the base peaks at m/e 270 corresponding to physcion-anthrone (4), which was obtained chemically by the thermal decomposition of the dimers. On treatment with HCl-HOAc, the dimers gave dianhydro-derivatives (6), which also yielded physcion-anthrone (4) on pyrolysis. The PMR spectra revealed the presence of following functional groups in the dimers; two phenolic, two enolic and two alcoholic OH groups, two Me groups, two OMe groups, four $-CH_2-$ groups, and four aromatic protons, and indicated that their planar structures (5) was identical

to that of phlegmacin B_1 which had been reported as the constituent of the fungus, *Cortinarius odorifer* Britz. (Agaricales) [3]. The strong Cotton effects in ORD have indicated that H and I are atropisomers caused by restricted rotation around biphenyl linkage and, at the same time, they must be diastereoisomers as indicated by differences in R_f . Attempts to racemize the chiral biphenyl bond resulted in the formation of complex mixtures probably arising from dehydration products. The result indicates that the both H and I are natural products and not artifacts formed during the isolation process by the racemization of one of the isomers.

Direct comparison using TLC and ORD was required to clarify the stereochemical relationship between the three phlegmacins, because their spectral data were almost identical. A sample of phlegmacin obtained from *C. odorifer* gave two spots on TLC, and chromatographic separation afforded two compounds which showed almost identical spectral data except ORD curves as in compounds H and I. The fungal compound showing a negative Cotton effect was defined as phlegmacin B_1 and the other compound showing a positive Cotton effect was named phlegmacin A_1 according to the nomenclature proposed by Steglich [4]. Although an atropisomeric

Table 1. 1H PMR data of dimeric hydroanthracenes obtained from the seedlings of *Cassia torosa*

	2-CH ₂	3-Me	3-OH	4-CH ₂	10-H	5-H	6-OMe	8-OH	9-OH	
Phlegmacin B ₂ (H)	2.84 br.s	1.47 s	2.05 s	3.07 br.s	6.93 s	6.63 s	3.73 s	9.90 s	15.98 s	
Phlegmacin A ₂ (I)	2.82 br.s	1.44 s	2.04 s	3.08 br.s	6.93 s	6.65 s	3.76 s	9.83 s	15.98 s	
					22%	22%				
Phlegmacin B ₂ (H)	2'-CH ₂	3'-Me	3'-OH	4'-CH ₂	5'-H	6'-OMe	7'-H	8'-OH	9'-OH	
	2.82 br.s	1.30 s	2.04 s	2.72 d	6.10 d	3.65 s	6.43 d	10.07 s	16.47 s	
				2.90 d	(J = 2.5)		(J = 2.5)			
Phlegmacin A ₂ (I)	2.73 br.s	1.28 s	2.04 s	2.63 d	6.13 d	3.64 s	6.46 d	10.08 s	16.45 s	
				2.85 d	(J = 2.5)		(J = 2.5)			
				(J = 17)						
Torosachryson (1)	2-CH ₂	3-Me	3-OH	4-CH ₂	5-H	6-MeO	7-H	8-OH	9-OH	10-H
	2.80 br.s	1.30 s	2.00 s	2.98 br.s	6.60 d	3.84 s	6.40 d	9.87 s	15.88 s	6.95 s
					(J = 2.5)		(J = 2.5)			
Physcion (3)	2-H	3-Me	4-H	5-H	6-OMe	7-H	8-OH	1-OH		
	7.04 br.s	2.45 s	7.57 br.s	7.32 d	3.92 s	6.60 d	12.05 s	12.26 s		
				(J = 2.5)		(J = 2.5)				
Anhydrophlegmacin-9,10-quinone B ₂ (J)	2-H	3-Me	4-H	5-H	6-OMe	8-OH	1-OH			
	7.05 br.s	2.47 s	7.59 br.s	7.51 s	3.64 s	11.98 s	12.43 s			
		31%	32%							
Anhydrophlegmacin-9,10-quinone A ₂ (K)	7.08 br.s	2.41 s	7.58 br.s	7.53 s	3.70 s	11.93 s	12.31 s			
Anhydrophlegmacin-9,10-quinone B ₂ (J)	2'-H	3'-Me	3'-OH	4'-CH ₂	5'-H	6'-OMe	7'-H	8'-OH	9'-OH	
	2.73 br.s	1.34 s	1.90 s	2.72 d	6.08 d	3.84 s	6.48 d	10.28 s	16.73 s	
				2.87 d	(J = 2.5)		(J = 2.5)			
				(J = 17)		24%	31%			
Anhydrophlegmacin-9,10-quinone A ₂ (K)	2.83 s	1.31 s	—	2.60 d	6.07 d	3.82 s	6.46 d	10.14 s	16.66 s	
				2.84 d	(J = 2.5)		(J = 2.5)			
				(J = 17)						

*Spectra were recorded in $CDCl_3$ at 100 MHz. Values are given in ppm (δ) relative to TMS as internal standard. Numbers in parenthesis denote coupling constants in Hz; s (singlet), br.s (broad singlet), d (doublet). Arrows and figures in % indicate enhancement in NOE experiments.

pair of 8'-methylphlegmacins had been isolated from *Corticarius percomis* Fr. [4], it was the first time that phlegmacin A₁ was so far isolated from *C. odorifer*. Direct comparison of phlegmacins A₁ and B₁ with our samples obtained from the seedlings clarified the stereochemical relationships among four phlegmacins. Although compound I showed the same R_f (0.25) as that of phlegmacin B₁, it showed a positive Cotton effect and a completely reverse ORD curve to that of phlegmacin B₁. The same relation was observed between compound H and phlegmacin A₁ (R_f 0.28). Thus it became clear compounds H and I are the enantiomers of phlegmacins B₁ and A₁ and should be named phlegmacin B₂ and phlegmacin A₂ respectively. PMR signals were assigned as shown in the Table by the aid of NOE experiments and also by the shifts of aromatic protons in the corresponding diacetates (6). Acetylation caused significant shifts of aromatic protons. One of the singlets (δ 6.6) and one of the doublets (δ 6.1) were shifted down field by 0.4 and 0.6 ppm, respectively, to be assigned to the aromatic protons *para* to the phenolic OH groups.

Due to the presence of the other chiral centres, the pair of isomers obtained from either *C. torosa* or *C. odorifer* are not only atropisomers but also diastereoisomers. Occurrence of two pairs of phlegmacins from two different sources indicates that the biosynthesis of monomeric moiety, torosachrynone (1), proceeds stereospecifically to produce (+)-torosachrynone in the higher plant and probably (–)-torosachrynone in the fungus, whereas the dimerization reactions in both living systems are not stereospecific.

From less polar chromatographic fractions, another pair of atropisomeric dimers, compounds J (R_f 0.62) and K (R_f 0.59), C₃₂H₂₆O₁₀, have been isolated. They showed positive Mg-acetate reaction indicating a quinone nature and almost identical spectral data to each other except ORD, in which the Cotton effects were completely opposite. PMR spectra revealed the presence of three phenolic, one enolic and one alcoholic OH groups, two Me groups, two –CH₂– groups, two MeO groups and five aromatic protons. Comparison of the PMR spectra with those of (1) and (3) as well as the phlegmacins (5) indicated that the quinonoid dimers contain both torosachrynone and physcion moieties. The presence of physcion moiety was verified by thermal decomposition yielding physcion instead of physcion-anthrone yielded in the thermal decomposition of phlegmacins (5). By the comparison of PMR spectra with those of phlegmacins linkages between two monomeric moieties was located between C-10' of torosachrynone moiety and C-7 of physcion moiety. This planar structure (7) is anhydrophlegmacin-9,10-quinone, the 8'-methyl ether of which has been isolated from *Corticarius percomis* Fr. [4] along with 8'-methylphlegmacin A₁ and B₁. If we assume that the quinonoid dimers (7) are derived from corresponding phlegmacins (5) by dehydration followed by oxidation, compound J and K should be named anhydrophlegmacin-9,10-quinone B₂ and A₂ respectively according to the signs of their Cotton effects. The PMR signals were assigned as shown in the Table by the comparison with those of phlegmacins (5). The signals due to the torosachrynone moieties, especially the signals of –CH₂– groups, are well accord with the corresponding signals of phlegmacins (6).

Recently, dimeric hydroanthracenes possessing similar structures were reported as the toxic constituents of *Karwinskia humboldtiana* Zucc. (Rhamnaceae) [5].

EXPERIMENTAL

Materials were the same as were used in previous report [2]. TLC was carried out with Si gel plate impregnated with *N*/2 oxalic acid developed with C₆H₆–EtOAc (4:1).

Extraction and isolation. C₆H₆ extracts of the seedlings (from 1 kg seeds) were chromatographed over Si gel using C₆H₆–EtOAc (4:1). The eluents were divided into 7 fractions by monitoring with TLC as described in previous paper [2]. Fr. 3 and 4 afforded crude compound J (50 mg) and compound K (80 mg) respectively. Crude compounds H (40 mg) and I (40 mg) were precipitated from the concentrated solution of Fr. 6 and 7 respectively.

Phlegmacin B₂ (Compound H) (5). Crude crystals obtained from Fr. 6 were further purified by Droplet Counter Current Chromatography (DCC) [6] using upper and lower phases of a solvent mixture, C₆H₆–CHCl₃–MeOH–H₂O (5:5:7:2). Resinous substances contained in the crude sample were removed by this process and after repeated recrystallization from Me₂CO compound H was obtained as dark brown rods, mp 198–200° (dec). UV $\lambda_{\text{dioxane}}^{\text{max}}$ nm (log ϵ): 233 (4.62), 278 (5.01), 320 (4.16), 333 (4.25); IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{–1}: 3400, 2920, 1630, 1615, 1585; MS 70 eV *m/e* (rel. int.): 574 (M⁺; 11.1), 556 (M⁺–H₂O; 12.2), 538 (M⁺–2H₂O; 84.2), 270 (100); High Resolution MS: Found; 538.1608, Calcd. for C₃₂H₂₆O₈ (M⁺–2H₂O); 538.1626; ORD [α] × 10^{–4} (nm): –50.7° (275), 0° (288), +7.5° (295), 0° (360).

Phlegmacin A₂ (Compound I) (5). Powder obtained from Fr. 7 was further purified with DCC using the same solvent system as for compound H. Compound I was finally obtained as dark brown powder, mp 194–5° (dec). UV, IR and MS were almost identical to those of compound H. High Resolution MS: Found; 538.1614; Calcd. for C₃₂H₂₆O₈ (M⁺–2H₂O); 538.1626; ORD [α] × 10^{–4} (nm): +48.6° (275), 0° (288), –8.1 (295), –0.2 (360).

8,8'-Diacylphlegmacin B₂ (7). Phlegmacin B₂ was acetylated with Ac₂O and C₆H₅N. After usual work up, the product was chromatographed over Si gel; A yellow fraction eluted with C₆H₆–Me₂CO (2:5) gave yellow needles mp 137°. IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{–1}: 3440, 2940, 1770, 1750, 1620. PMR (100 MHz; CDCl₃): 1.23, 1.43 (3H × 2, s × 2, Me × 2), 2.20, 2.40 (3H × 2, s × 2, Ac × 2), 2.64, 2.84 (1H × 2, d × 2, J = 17, CH₂), 2.68 (2H, s, CH₂), 2.82 (2H, s, CH₂), 3.12 (2H, s, CH₂), 3.60, 3.80 (3H × 2, s × 2, OMe × 2), 6.42, 6.70 (1H × 2, d × 2, J = 2.5, ArH × 2), 7.02, 7.08 (1H × 2, s × 2, ArH × 2), 14.50, 15.00 (1H × 2, s × 2, OH × 2).

8,8'-Diacylphlegmacin A₂ (6). This acetate was obtained as yellow powder mp 178°. IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{–1}: 3340, 2940, 1770, 1750, 1620. PMR (100 MHz; CDCl₃): 1.23, 1.43 (3H × 2, s × 2, Me × 2), 2.20, 2.40 (3H × 2, s × 2, Ac × 2), 2.54, 2.74 (1H × 2, d × 2, J = 17, CH₂), 2.80 (4H, s, CH₂ × 2), 3.10 (2H, s, CH₂), 3.65, 3.85 (3H × 2, s × 2, OMe × 2), 6.50, 6.72 (1H × 2, d × 2, J = 2.5, ArH × 2), 7.03, 7.08 (1H × 2, s × 2, ArH × 2), 14.56, 15.07 (1H × 2, s × 2, OH × 2).

Physcion-anthrone (4) from phlegmacin A₂ and B₂. Phlegmacin B₂ (20 mg) was heated *in vacuo* at 250°. Colourless needles sublimed in a tube showed M⁺ 270 in MS and were identified as physcion-anthrone (3) by the comparison of IR with that of authentic sample. Physcion-anthrone was further converted into physcion (3) with KOH–EtOH and identified by mmp. Phlegmacin A₂ (20 mg) also gave physcion-anthrone (5 mg) under the same condition.

Dianhydrophlegmacin A₂ and B₂. Phlegmacin B₂ (20 mg), AcOH (1 ml) and conc HCl (0.01 ml) were heated at 80° for 40 min. Resulted black soln was diluted with H₂O and extracted with EtOAc. The extracts was chromatographed over Si gel and dianhydrophlegmacin B₂ was obtained as colourless needles (5 mg) mp > 270° (from Me₂CO). MS 70 eV *m/e* (rel. int.): 538 (M⁺; 85.0), 270 (100), 255 (10.0), 242 (9.0), 227 (9.3). Pyrolysis of dianhydrophlegmacin B₂ (15 mg) afforded physcion-anthrone (4) (4 mg). Phlegmacin A₂ (20 mg) also yielded dianhydrophlegmacin A₂ (5 mg) mp > 270°, which showed identical IR and MS spectra to those of dianhydrophlegmacin B₂.

Anhydrophlegmacin-9,10-quinone B₂ (Compound J) (7). Crude crystals obtained from Fr. 3 were repeatedly recrystallized from

MeOH to give dark red needles, mp 192–3°. UV $\lambda_{\text{dioxane}}^{\text{max}}$ nm (log ϵ): 218 (4.63), 268 (4.71), 300 (4.12), 406 (4.16); IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3500, 3400, 2980, 1674, 1640, 1625, 1595; MS 70 eV m/e : 570 (M^+), 552 ($M^+ - H_2O$), 284 (base peak). High Resolution MS: Found; 552.1416, Calcd. for $C_{32}H_{24}O_9(M^+ - H_2O)$; 552.1419. ORD $[\alpha] \times 10^{-3}$ (nm): -24.8° (278), 0° (293), 0° (318).

Anhydrophlegmacin-9,10-quinone A₂ (Compound K) (7). Dark red powder obtained from Fr. 4 was repeatedly chromatographed over Si gel to give compound K as dark red powder. UV, IR and MS were identical to those of compound J. ORD $[\alpha] \times 10^{-3}$ (nm): +27.1° (275), 0° (288), +9.2° (292), 0° (317).

Physcion (3) from anhydrophlegmacin-9,10-quinone B₂ (7). Pyrolysis of anhydrophlegmacin-9,10-quinone B₂ (10 mg) yielded yellow crystals, which were purified by column chromatography over Si gel to give physcion (3 mg) mp 205°.

Phlegmacin B₁ and phlegmacin A₁ from Cortinarius odorifer. Crude phlegmacins (300 mg) obtained from *C. odorifer* and generously supplied by Dr. W. Steglich were chromatographed over Si gel with C₆H₆-AcOEt (4:1). First yellow band gave dark brown crude crystals of phlegmacin A₁ (100 mg) and second yellow band gave crude phlegmacin B₁ (120 mg). Phlegmacin A₁ was obtained as dark brown needles mp 195° from Me₂CO. UV $\lambda_{\text{dioxane}}^{\text{max}}$ nm (log ϵ): 233 (4.67), 278 (4.93), 320 (4.17), 330 (4.09), 400 (4.33). IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3400, 3930, 1630, 1610, 1590. PMR (60 MHz; CDCl₃): 1.28, 1.43 (3H \times 2, $s \times 2$, Me \times 2), 2.82* (6H, $br.s.$, CH₂ \times 3), 3.08 (2H, $br.s.$, CH₂), 3.62, 3.73 (3H \times 2, $s \times 2$, OMe \times 2), 6.11, 1.46 (1H \times 2, $d \times 2$, $J = 2.5$ Hz, ArH \times 2), 6.70, 6.98 (1H \times 2, $s \times 2$, ArH \times 2), 9.98, 10.13, 16.10, 16.60 (1H \times 4, $s \times 4$, OH \times 4). ORD $[\alpha] \times 10^{-4}$ (nm): +27.1°

(275), 0° (288), -10.9° (294), 1.13° (360). Phlegmacin B₁ was obtained as dark brown needles mp 196° from Me₂CO. UV, IR and MS are identical to those of phlegmacin A₁. PMR (60 MHz; CDCl₃): 1.30, 1.46 (3H \times 2, $s \times 2$, Me \times 2), 2.82* (4H, s , CH₂ \times 2), 2.73, 3.04 (2H \times 2, $s \times 2$, CH₂ \times 2), 3.73, 3.64 (3H \times 2, $s \times 2$, OMe \times 2), 6.14, 6.47 (1H \times 2, $d \times 2$, $J = 2.5$ Hz, ArH \times 2), 6.67, 6.97 (1H \times 2, $s \times 2$, ArH \times 2), 9.94, 10.12, 16.12, 16.59 (1H \times 4, $s \times 4$, OH \times 4). High Resolution MS: Phlegmacin A₁ Found; 538.1624; Phlegmacin B₁ Found; 538.1644, Calcd. for $C_{32}H_{26}O_8(M^+ - 2H_2O)$; 538.1626.

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* Due to lower resolution these signals appeared as broad singlets.